High pressure phase transition and strength estimate in polycrystalline alumina during laser-driven shock compression

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Abstract: Alumina (Al₂O₃) is an important ceramic material notable for its compressive strength and hardness. It represents one of the major oxide components of the Earth's mantle. Static compression experiments have reported evidence for phase transformations from the trigonal α -corundum phase to the orthorhombic Rh₂O₃(II)-type structure at ~90 GPa, and then to the post-perovskite structure at ~130 GPa, but these phases have yet to be directly observed under shock compression. In this work, we describe laser-driven shock compression experiments on polycrystalline alumina conducted at the Matter in Extreme Conditions endstation of the Linac Coherent Light Source. Ultrafast X-ray pulses (50 fs, 10¹² photons/pulse) were used to probe the atomic-level response at different times during shock propagation and subsequent pressure release. At 107±8 GPa on the Hugoniot, we observe diffraction peaks that match the orthorhombic Rh₂O₃(II) phase with a density of 5.16±0.03 g/cm³. Upon unloading, the material transforms back to the α -corundum structure. Upon release to ambient pressure, densities are lower than predicted assuming isentropic release, indicating additional lattice expansion due to plastic work heating. Using temperature values calculated from density measurements, we provide an estimate of alumina's strength on release from shock compression.

1. Introduction

Alumina (α -Al₂O₃) is a model material in ceramics and an important oxide component of the Earth. Due to its hardness and high compressive strength, it has a wide variety of industrial uses including in abrasives, insulators, windows and spark plugs (Doremus, 2008, pp. 1-26). In static high-pressure experiments using diamond anvil cells, Cr³⁺ fluorescence from the Al₂O₃ variety ruby has long been used as a pressure calibrant (Mao et al, 1978, pp. 3276-3283).

Al₂O₃ constitutes up to 6% of the Earth's lower mantle by weight, and is expected to be almost entirely incorporated in the perovskite and post-perovskite phases of MgSiO₃. (Irifune, 1994, pp.131-133) Alumina's crystal structure under high pressures and temperatures is of interest due to its potential effects on the thermodynamics of phase transformations in MgSiO₃ under lower mantle conditions, particularly near the core-mantle boundary (136 GPa) (Tsuchiya, Tsuchiya and Wentzcovitch, 2005, p.020103).

Static compression studies have probed the structure of alumina using powder X-ray diffraction (XRD), and identified a phase transformation from the trigonal α -corundum phase (space group *R-3c*) to an orthorhombic phase. This transformation was first reported by Funamori & Jeanloz (1997, pp.1109-1111) in polycrystalline ruby at a pressure of 100±10 GPa and ~1000 K. The phase transformation was later characterized in more detail by Lin et al (2004, pp.389-393), who found the transition to occur on compression to 109 GPa at 300 K, and to 96 GPa at ~1000 K. The new phase persisted down to 85 GPa upon decompression. Rietveld

full-profile analysis showed that the diffraction peak shapes matched the Rh₂O₃(II)-type structure (*Pbcn*) rather than a perovskite structure (*Pbnm*). These findings are consistent with computational studies, which have found that the Rh₂O₃(II) phase becomes energetically favorable at pressures similar to those reported by Lin et al (2004, pp.389-393), as shown in Fig. 1. The perovskite phase is not expected to be stable in any pressure range at ambient temperature, but has been hypothesized to have a stability field around 100 GPa at temperatures above 2400 K (Oganov & Ono, 2005, pp.10828-10831). Static and computational studies have also found a second phase transition to the orthorhombic post-perovskite phase (*Cmcm*) at ~130 GPa (Tsuchiya, Tsuchiya and Wentzcovitch, 2005, p.020103; Oganov and Ono, 2005, pp.10828-10831). The α -corundum phase may be metastable up to pressures above both these phase boundaries; one static experiment found ruby to remain as α -corundum up to 175 GPa at room temperature (Jephcoat et al, 1988, pp.115-121).



Fig. 1: High-pressure phase diagram of Al₂O₃. The stability of the Rh₂O₃(II) phase was observed by Lin et al, (2004, pp.389-393) in static compression experiments on both compression (red triangles) and decompression (blue triangles), demonstrating agreement with theoretical phase boundaries computed by Tsuchiya, Tsuchiya and Wentzcovitch (2005, p.020103). Black circles represent α -corundum observed by Lin et al, (2004, pp.389-393). The calculated shock Hugoniot curve from SESAME EOS #7410 (single phase EOS) is shown in green (Lyon and Johnson, 1992). The melting curve was measured by Shen and Lazor (1995, pp.17699-17713) under static compression.

Shock compression is another technique used to study the behavior of materials at extreme pressures and temperatures. Mashimo et al (2000, pp.2021-2024) studied *a*-cut sapphire (single crystal Al_2O_3) under shock compression with a light gas gun using the inclined mirror method and reported a -3% volume discontinuity at 79 GPa and 1130 K on the shock Hugoniot. While this is suggestive of a volume collapse due to a phase transition, the crystal structure was not measured directly. Later experiments performed on *c*-cut sapphire by Reinhart, Chhabildas

and Vogler (2006, pp.655-669) using a two-stage gas gun found no evidence of a phase transition up to 170 GPa in shock velocity - particle velocity measurements. The difference in results when compressing along different orientations of a single crystal raises the question of how a polycrystalline sample with no preferred orientation might respond under shock compression.

Recent developments in laser compression paired with *in situ* X-ray diffraction (XRD) now allow us to probe crystal structure directly during shock propagation (Duffy and Smith, 2019, p.23; Glenzer et al, 2016, p.092001). This has become an important method to study phase diagrams at much higher pressures than can be obtained with static methods (Wicks et al, 2018, p.5864; Lazicki et al, 2021, pp.532-535). However, it remains an open question whether phase transformations observed under static conditions are always accessible under the short timescales of shock compression (Langenhorst and Deutsch, 2021, pp.31-36; Pavlovskii and Drakin, 1966, p.116). Comparing results between static and dynamic experiments at similar pressures can help characterize kinetics, thermodynamics, and mechanisms of structural transitions in materials (McBride et al., 2019, pp.89-94).

In situ XRD also provides new ways to constrain material strength at high pressures. This has been achieved by determining the amount of anisotropic elastic strain, i.e., difference in elastic strain between the loading and transverse directions. (Turneaure and Gupta, 2011, p.123510) Another way to infer material strength is by obtaining information about the path taken upon pressure release from shock. In particular, excess temperature during release relative to the expected rate of thermoelastic cooling is indicative of plastic-work heating, which directly results from dynamic strength (Heighway et al, 2019, p.245501). Determining whether a material maintains or loses its strength during unloading is of key interest in understanding its deformation.

The strength of alumina is of broad interest due to its widespread applications. Notably, polycrystalline alumina has been found to undergo an increase in and retention of high dynamic strength (~20 GPa) under shock to up to 100 GPa, unlike single-crystal sapphire which undergoes a loss of strength at lower pressures (Kleiser, Chhabildas and Reinhart, 2011, pp.473-479; Ahrens, Gust and Royce, 1968, pp.4610-4616). Measurements of alumina's strength can also inform predictions of strength for other sesquioxides including Fe₂O₃, Rh₂O₃ and Cr₂O₃, which adopt the α -corundum structure at ambient conditions (Lin et al, 2004, pp.389-393).

In this study, we report measurements of alumina's crystal structure during shock compression, reaching up to 107±8 GPa on the Hugoniot. Our study provides the first direct observation of a high-pressure phase transition in shock-compressed alumina. Using the density of alumina upon full release to ambient pressure, we estimate the plastic work heating and the corresponding strength of alumina under dynamic unloading. These results can serve as a reference for the behavior of other ceramic materials and understanding the physics of material deformation at nanosecond time scales.

2. Experimental Methods

Samples of 99.6% pure polycrystalline alumina with no porosity and no preferred orientation were purchased from Valley Design Corp. X-ray diffraction patterns taken prior to compression confirm the sample identity and purity.

Laser-driven shock compression experiments were performed at the Matter in Extreme Conditions endstation of the Linac Coherent Light Source (LCLS). LCLS is an X-ray Free Electron Laser (XFEL) that delivers coherent, vertically polarized, ~50 fs X-ray pulses, with ~ 10^{12} photons/pulse. Each target package consisted of a 50-µm thick polyimide ablator layer, a 0.2-µm aluminum or 1-µm gold reflective coating, and 60 µm of polycrystalline alumina (Fig.

2a). The target was shock compressed by focusing a 10-13 ns near flat top drive laser pulse, delivered by either one or two beams of a 527-nm Nd-YAG glass laser, onto the polyimide ablator surface. Waveplates were tuned between 45-53° to create a focal spot size of ~250 μ m on the ablator surface and reach the desired sample stress state (Brown et al, 2017, p.105113). Two time series of five shots each were conducted on alumina at nominal pressures of ~50 and ~110 GPa using one or two overlapping drive beams, respectively. Targets with a gold reflective coating were used for the single-beam shots (runs 199 to 209), while those with an aluminum coating were used for the dual-beam shots (runs 252 to 273). The X-ray beam's spot diameter was ~20 μ m in the single-beam shots and ~80 μ m in the dual-beam shots. VISAR velocimetry data were collected, but almost no signal was observed with this target setup due to the opacity of the sample.



Fig. 2: Target setup and CSPAD detectors. (A) Schematic of the experimental setup (not to scale). One or two beams of the 527-nm Nd:Glass laser were focused onto a ~250- μ m spot on the ablator surface. Targets used for the single-beam shots had a gold reflective coating, while those used for the dual-beam shots had an aluminum coating. A 50-fs pulse of the LCLS XFEL was directed onto the sample at a 24° angle of incidence and focused to a spot size of 20 or 80 μ m. Scattered x-rays were detected on large area Cornell-SLAC Hybrid Pixel Array Detectors (CSPAD) (Herrmann et al., 2013, pp.550-553) (B) Cartesian projection using the HEXRD analysis software showing the positions of the CSPADs and the angular range of data collection. The diffraction pattern shown was collected in run 255.

The crystal structure of alumina was probed during shock and release using coherent, vertically polarized, 8.02 keV, ~50 fs X-ray pulses produced by the XFEL. The X-ray beam entered the target package at a 24° angle of incidence. Runs 232 and earlier used an X-ray spot diameter of 20 μ m, while in runs 233 and later it was 80 μ m. X-ray diffraction (XRD) patterns were recorded by five Cornell-SLAC Pixel Array Detectors (CSPAD) with 16-71° angular 20 coverage (Fig. 2b). X-ray delays with respect to the drive laser ranged from 7 to 20 ns, probing the sample at various times relative to shock propagation and release. Detector positions and orientations were determined using diffraction patterns from a CeO₂ calibrant, and cross-validated with diffraction from the uncompressed Al₂O₃ starting material.

The HEXRD analysis software was used to convert XRD patterns to polar coordinates and perform Lorentz-polarization and pixel solid angle corrections on the data (Boyce and Bernier, 2013).

Azimuthally averaged diffraction patterns were fit to a pseudo-Voigt peak shape using the Pawley method with the powder diffraction tool in Fityk (Wojdyr, 2010). This fitting was used to determine lattice parameters and density during and after shock.

Pressure was determined using the linear U_s - u_p (shock velocity - particle velocity) relation of Al₂O₃ experimentally determined in several studies up to 300 GPa (Ahrens, Gust and Royce, 1968, pp.4610-4616; Barker and Hollenbach, 1970, pp.4208-4226; Erskine, 1993; Graham and Brooks, 1971, pp.2311-2330; Kleiser, Chhabildas and Reinhart, 2011, pp.473-479; Mashimo, Hanaoka and Nagayama, 1988, pp.327-336; Marsh, 1980; Reinhart, Chhabildas and Vogler, 2006, pp.655-669). These, together with the following Rankine-Hugoniot equations for conservation of mass (Eqn. 1) and momentum (Eqn. 2), were used to calculate pressure from the densities measured during shock, via Eqn. 3.

$$\frac{\rho_0}{\rho} = \frac{U_s - u_p}{U_s} (1)$$

$$P = \rho_0 U_s u_p (2)$$

$$P = \rho u_p (U_s - u_p) (3)$$

Most of the previous Hugoniot studies were conducted on single-crystal sapphire. Given the large strength of Al_2O_3 , the Hugoniot curve could differ between the single crystal and polycrystalline forms. Mashimo, Hanaoka and Nagayama (1988, pp.327-336) and Ahrens, Gust and Royce (1968, pp. 4610-4616) measured the Hugoniot of both single crystal and polycrystalline Al_2O_3 , and determined that while there is a difference in the Hugoniot Elastic Limit, the deformational Hugoniot curve is nearly identical for non-porous samples, such as used here.

The HYADES hydrocode was used to perform 1-dimensional simulations of shock propagation and release for all shots (Larsen and Lane, 1994, pp.179-186). The kapton ablator was modeled using the SESAME #7770 equation of state (EOS) table for parylene, while alumina was modeled using SESAME EOS #7410. The known relative laser power as a function of time for each shot, shown in Fig. S3 in the Supplementary Materials, was applied to the kapton surface, yielding pressures of ~50 GPa or ~110 GPa in alumina. Example simulations of runs 269 and 205 are shown in Fig. 4.

3. Results

We measured two time series of X-ray images, revealing the changes in crystal structure of polycrystalline alumina through shock compression and subsequent pressure release (see Table 2 and Fig. 3). Targets were shocked to a nominal pressure of \sim 50 GPa or \sim 110 GPa using one or two beams of the MEC drive laser focused onto a \sim 250-µm spot on the kapton ablator,

and X-ray images were collected at different times relative to the drive laser pulse (defined as t=0). Diffraction from ambient Al₂O₃ was observed in many of the images, and several include diffraction from both shocked and decompressed material. Timings of partial release from the kapton-alumina interface were deduced from drops in density observed in the compressed material over time. The timings of shock breakout from the free surface of alumina were, in the single-beam shots, determined based on presence of diffraction peaks from the unshocked starting material, which disappear once the shock wave reaches the free surface (Table 2).

In the single-beam shots (shocked to ~50 GPa), alumina remained in the α -corundum phase both on shock and release, with no apparent change in texture (Fig. 3a and Fig. S1a). These targets included a gold reflective coating whose diffraction is also observed during compression. The *c/a* ratio of alumina was observed to be 2.72-2.74 for all shots in this sequence (compression and release) except at 10 ns where a value of 2.69 was observed. Release from the shocked state begins between 16 and 18 ns, evidenced by a sharp increase in *d*-spacing between these X-ray images. At 20 ns, the latest time sampled, the material remains partially compressed. At the earlier times of 10 and 14 ns, a spotty diffraction signal with two-theta values intermediate between those of the ambient and fully compressed states is observed. This could arise from elastically compressed material, but future work is needed to better understand these observations.

In the dual-beam shots, alumina was shocked to ~110 GPa. Diffraction at 9 ns (run 269) does not include any peaks that could be indexed to the shocked α -corundum phase. We instead observe evidence for a new phase that shows a change in the azimuthal intensity distribution around the Debye-Scherrer cones. The diffraction patterns match both the Rh₂O₃(II) phase and the perovskite phase with a density of 5.16±0.03 g/cm³ for each (see Fig. 3c). We assign the

Rh₂O₃(II) phase based primarily on previous static and theoretical results. Peak fitting with the Pawley method yields a better fit with lower reduced χ^2 for the Rh₂O₃(II) phase, but due to the presence of overlapping peaks, it is difficult to draw concrete conclusions about the goodness of fit (see Table 1, Table S2 and Fig. S4). We also observe three additional peaks that match the Rh₂O₃(II) phase at a density of 4.80±0.01 g/cm³ on decompression, indicated in Fig. 3c with red circles. These appear as spots rather than lines, suggesting further development of preferred orientation as the sample begins to unload.

In diffraction patterns taken at later times of 10 and 11 ns (runs 252 and 270), peaks consistent with the α -corundum phase are observed and the corresponding densities are 4.36±0.04 g/cm³ and 4.35±0.01 g/cm³, consistent with back-transformation on release from the shocked state. Similar to the 9-ns timing, these diffraction lines exhibit a change in texture relative to the starting material, suggesting a transition to the new phase and back to α -corundum. By 13 ns (run 273), the material has fully returned to ambient pressure and displays a uniform texture, suggesting a loss of texture with release which may be due to fracturing and break-up of grains on decompression, creating a more uniform powder.

For the dual-beam shots, the X-ray spot size was increased from 20 μ m to 80 μ m in order to increase the number of grains sampled by the X-ray beam and smooth out the texture of the polycrystalline sample. As the laser spot size remained at ~250 μ m, the larger sampled region likely resulted in diffraction signal from laterally released material from the edges of the shocked region. We attribute diffraction from material whose densities are considerably lower than that of the starting material to this laterally released material. The wider X-ray focus also resulted in diffraction from ambient, starting material outside the shocked region, including after shock breakout.

Structure	a (Å)	b (Å)	<i>c</i> (Å)	ρ (g/cm ³)	Reduced χ^2	R ²
Rh ₂ O ₃ (II) (Pbcn)	6.457±0.004	4.448±0.006	4.564±0.001	5.16±0.01	0.702	0.9961
Perovskite (Pbnm)	6.422±0.003	4.545±0.001	4.499±0.020	5.16±0.03	2.709	0.9782

Table 1: Comparison of Pawley fits to the $Rh_2O_3(II)$ and perovskite structures in the shocked phase of run 269 (9 ns X-ray delay). Both structures are good fits to the observed diffraction peaks, but $Rh_2O_3(II)$ has a lower reduced χ^2 and higher R^2 . Since our data include some overlapping peaks with unclear distinctions, and we refine three independent cell parameters, our error bars likely underestimate the true uncertainty in lattice parameters and density.

Shock breakout time (ns)	Partial release from interface (ns)	Run #	Drive energy (J)	X-ray delay (ns)	Structure	State	ρ (g/cm ³)	P (GPa)	
Single Beam									
16-18	16-18	209	17.4	10	a-cor	Shocked	4.546±0.020	42±4	
		205	16.6	14	a-cor	Shocked	4.618±0.009	43±4	
		202	16.9	16	a-cor	Shocked	4.541±0.018	44±4	
		200	15.8	18	a-cor	Releasing	4.022±0.007	-	
		199	14.7	20	a-cor	Releasing	4.006±0.006	-	
Dual Beam	l				•	·			
11-13	9-10	255	32.5	7	a-cor	Ambient	3.989±0.000	0.0001	
		269	30.1	9	Rh ₂ O ₃ (II)	Shocked	5.164±0.036	107±8	
					Rh ₂ O ₃ (II)	Releasing	4.805±0.012	-	
					a-cor	Released	3.801±0.003	~0	
		252 31	31.9	10	a-cor	Releasing	4.359±0.046	-	
					a-cor	Released	3.859±0.004	~0	
		270	30.3	11	a-cor	Releasing	4.346±0.013	-	
					a-cor	Released	3.853±0.002	~0	
		273	34.2	13	a-cor	Released	3.837±0.003	~0	

Table 2: Summary of X-ray diffraction data. Some runs include diffraction from multiple phases. In run 269, which includes the new phase, the densities and pressures shown assume the material is in the

 $Rh_2O_3(II)$ phase. Diffraction from the starting material is observed in several runs but is not included in this table, with the exception of run 255, in which it is the only phase present. In the single-beam shots, the timing of shock breakout from the rear free surface of alumina was determined based on the presence or absence of diffraction peaks from unshocked starting material at different X-ray timings. In the dual-beam shots, shock breakout time was determined from observation of only released material at 13 ns. Timings of partial pressure release from the kapton-alumina interface were estimated based on sharp drops in densities over time.

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Fig. 3: X-ray diffraction patterns of shocked alumina. (A) Diffraction pattern of run 205, taken at 14 ns. Dashed vertical black and brown lines indicate peak positions of the starting material and shocked

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 Al_2O_3 respectively. Miller indices of the diffracting planes are shown. An oversaturated color scale is used to highlight details in darker areas, as the intensity ratio between the brightest and darkest peaks is large. (B) Azimuthally averaged diffraction patterns of the single-beam shots. Lineouts are ordered by X-ray timing. Peaks that correspond to compressed (shocked or releasing) α -corundum are labeled with brown ticks, while diffraction peaks from the ambient starting material are labeled with blue ticks. Diffraction from the 1 µm-thick gold reflective coating is labeled with yellow ticks. Diffraction from the gold reflective coating is labeled with yellow ticks. Gold diffraction is not visible at 18 and 20 ns (runs 200 and 199). This may be due to melting during release from shock, as the pressure is lowered while temperature remains high. Based on the expected shock pressure, this would only happen if the gold undergoes a nonisentropic release and heats up during release. Alternatively, it is possible that the gold peaks are present but fully overlap with the Al2O3 (110) and (113) peaks. If this is the case, the corresponding density would be 22.94 g/cm³, consistent with gold during decompression (Briggs et al, 2019, p.045701). (C) Diffraction pattern of run 269, taken at 9 ns. Black and dark blue dotted lines indicate the calculated peak positions of α -corundum from the starting material and material released from shock, respectively. Dashed red lines indicate peak positions fit to the Rh₂O₃(II) structure. Peaks from the starting material and released material in the α -corundum phase show up in closely spaced pairs so only one label is provided for each pair. The red circles with Miller indices highlight spots of lower density Rh₂O₃(II) phase. As in (A), the data are not oversaturated, but an oversaturated color scale is used to highlight important details. (**D**) Diffraction patterns of the dual-beam shots. Peaks assigned to the $Rh_2O_3(II)$ structure in the compressed state are labeled with red ticks, while those corresponding to a-corundum are labeled with blue, cyan and brown ticks depending on the compression state. Diffraction from the aluminum reflective coating is labeled with a gray tick. In the 7-ns profile, broad peaks at $\sim 41^{\circ}$ and $\sim 48^{\circ}$ are consistent with diffraction from the amorphous polyimide ablator (Katagiri et al, 2022, p.054103). See Fig. S1 in the Supplemental Materials for two-dimensional diffraction images from all runs.



Fig. 4: Pressure-density Hugoniot curve. Hugoniot data of single-crystal sapphire and polycrystalline alumina plotted in pressure-density spaces (Ahrens, Gust and Royce, 1968, pp.4610-4616; Barker and Hollenbach, 1970, pp.4208-4226; Erskine, 1993; Graham and Brooks, 1971, pp.2311-2330; Kleiser,

Chhabildas and Reinhart, 2011, pp.473-479; Mashimo, Hanaoka and Nagavama, 1988, pp.327-336; Marsh, 1980; Reinhart, Chhabildas and Vogler, 2006, pp.655-669). Data from studies on c-cut sapphire are shown in black, a-cut in blue, m-cut in green, and polycrystalline alumina in red. The discontinuity in the Hugoniot indicates the Hugoniot elastic limit (HEL) of single crystal sapphire. The HEL of polycrystalline alumina is lower, but the deformational Hugoniot curve for alumina with no porosity has been found to be very similar to that of sapphire (Mashimo et al 2000, pp.2021-2024, Ahrens, Gust and Royce, 1968, pp.4610-4616). The deformational shock Hugoniot curve, shown in dark green, was obtained by linearly fitting the Hugoniot data above the HEL in U_s-u_p space and using the Rankine-Hugoniot conservation equations to convert to pressure-density space. The dark green dashed lines indicate the error bounds of the fit. A 3% volume discontinuity associated with the α-corundum to Rh₂O₃(II) phase transition has been observed in static studies and continuum measurements, but was not taken into account in this fit (Lin et al, 2004, pp.389-393, Mashimo et al, 2000, pp.2021-2024). The blue asterisk indicates the pressure-density state of the starting material. Filled circles with error bars show the state measured during shock, which is constrained by the Hugoniot. Vertical dashed lines show the densities of the partially released states. Material that has released from compression with a two-beam drive laser and is near or at ambient pressure is labeled with a pink X.

4. Discussion

High Pressure Phase Transition

Our study provides direct evidence of a phase transition in alumina under shock compression at 107±8 GPa. This is consistent with static compression results and computational studies (Funamori and Jeanloz, 1997, pp.1109-1111; Lin et al, 2004, pp.389-393; Oganov and Ono, 2005, p.116; Thomson, Wentzcovitch and Bukowinski 1996, pp.1880-1882; Tsuchiya, Tsuchiya and Wentzcovitch, 2005, p.020103; Umemoto and Wentzcovitch, 2008, pp.6526-6530).

The diffraction patterns of the high-pressure phase fit both the $Rh_2O_3(II)$ (*Pbcn*) and perovskite (*Pbnm*) structures. Given the known Hugoniot pressure-density relation, the densities calculated for both the $Rh_2O_3(II)$ and perovskite structures (5.16 ± 0.03 g/cm³) are consistent with the expected shock pressure of ~110 GPa. Due to the similarity in diffraction patterns between these two phases, it was difficult to differentiate between them, but results from computational free energy studies (Oganov and Ono, 2005, pp.10828-10831) show that the perovskite structure is not likely to be stable at the expected temperatures in our experiment (based on SESAME EOS #7410). Rietveld full-profile refinements on alumina at comparable P-T conditions in diamond cell experiments also support the stability of the $Rh_2O_3(II)$ phase (Lin et al, 2004, pp.389-393)). These considerations lead us to believe that our shocked material most likely transformed to the Rh₂O₃(II) phase.

The $Rh_2O_3(II)$ phase also yielded a slightly better fit, with a lower reduced χ^2 than perovskite (Table 1). However, there is some ambiguity in hkl assignment where the peaks are highly textured and overlapping, so this distinction is not clear-cut.

Static compression experiments that report observation of the $Rh_2O_3(II)$ phase find a wide mixed-phase region in which both α -corundum and $Rh_2O_3(II)$ are observed (Funamori and Jeanloz, 1997, pp.1109-1111; Lin et al, 2004, pp.389-393). In our measurement at 107±8 GPa, we observe the high pressure orthorhombic structure with no trace of a shocked α -corundum phase. This may be due to the phase transition having a high kinetic barrier, leading to a slow transition and incomplete conversion to the high pressure phase in diamond anvil cell studies. With a uniaxial shock, non-hydrostaticity of the compression and the presence of shear stresses may make it easier to cross this barrier and fully convert to the new phase.

Temperature and Plastic Work

Dynamic release from the shock state is usually assumed to be a thermodynamically reversible process and hence isentropic. This assumption neglects possible irreversible heating associated with plastic deformation. The release rate in these experiments is so rapid (~ns) that the sample may retain dynamic strength during decompression. Such strength may result in substantial plastic-work heating and cause the release path to deviate markedly from the isentrope (Heighway et al, 2019, pp.532-535; Swift et al, 2007, p.054122). Our data on decompression from ~110 GPa enable us to place constraints on this heating to gauge the typical dynamic strength on unloading.

As illustrated in Fig. S5 and described in the Supplemental Material, in the X-ray images taken at 9, 10 and 11 ns we sample what is most likely laterally released material—shocked material that has equilibrated to the surrounding ambient material over time. We find that this material has a density lower even than that of unshocked material. This indicates that alumina has thermally expanded due to heating during the release.

The shock wave breaks out from alumina's free rear surface between 11 and 13 ns, evidenced by a loss of peaks from the ambient unshocked material. After shock breakout, a backward-traveling release wave from the rear surface drops the normal stress σ_L to zero. At 13 ns, we observe only peaks from released material suggesting that by this time, the backward-traveling release wave has overlapped with the forward-traveling release wave from the alumina-kapton interface, which may draw the sample into tension ($\sigma_L < 0$). Heating during release has been studied previously in metals (Heighway et al, 2019, pp.532-535), in which the interaction of release waves usually causes a sharp decrease in density far greater than that which could be caused by heating. However, in alumina, at 13 ns we actually measure a released density similar to those observed at earlier times. The lack of ellipticity in our diffraction rings indicates that the material does not experience significant tensile stress along the unloading axis. This suggests that the interaction of release waves causes material break-up rather than lattice expansion.

The temperature during dynamic release can be calculated from our measured densities and the known temperature-dependence of thermal expansivity of alumina (Fig. S2) (Watchman, Scuderi and Cleek, 1962, pp.319-323; Hahn, 1978, pp. 191-201; Schauer, 1964, pp.1857-1864; Amatuni and Shevchenko, 1976, pp.1256-1260; Yates, Cooper and Pojur, 1972, p.1046; Fitzer and Weisenburger, 1972, pp.25-35; Ebert and Tingwald, 1936, p.471; Mauer and Bolz, 1958; Campbell and Grain, 1961, pp.244-256; Kompan et al, 1999, pp.776-781; Kostanovskiy, et al, 2017, p.012317). At 9, 10 and 11 ns, our HYADES simulations suggest that the low-density signal we observe in our diffraction data can only be from laterally released material near the edge of the drive spot. If this material is under any nonzero, residual stress (due to the release being incomplete), we expect that stress to be positive. This means that the released material must be at least as hot as the zero-pressure thermal expansion data would have us believe; in two specimens of the same density, the one under greater pressure must have the greater temperature. Conversely, we expect that the later-time signal at 13 ns is from fully released material after the backward-facing release wave has overlapped with both front-facing and lateral waves. If there is any stress on the material due to the crossing of the release waves, we expect it to be tensile, although the lack of evidence for anisotropic strain suggests that this is likely not the case.

One confounding factor in interpreting these measurements is that there are small variations in laser pulse shape between runs, so the achieved shock pressure may vary (see Fig. S3). We are also sampling different portions of the material at different times; if there is any spatial variability in the stress state, different portions of the material may possess different strengths. Stronger material tends to release faster and release more heat, which may explain why our 9-ns run shows a higher nominal temperature than those taken at later times. More sophisticated simulations would be required to gain a better sense of the stress and temperature inhomogeneity throughout the sample.

In Fig. 5, we show both the Hugoniot and the isentropic release path emanating from the 107 GPa shock state in pressure-temperature space, calculated using a HYADES hydrodynamic simulation under SESAME EOS #7410. We predict that after being elevated to a temperature of around 1370 K by a 107 GPa shock, alumina would cool to 930 K were it to then release

isentropically. However, based on our diffraction measurements of released material, we find that the post-release temperature is at least 1580 K, far higher than that predicted by isentropic release using the SESAME EOS (and, indeed, higher even than the shock temperature itself). We suggest the excess heat comes from plastic-work heating, in which heat is generated by frictional forces acting during shear plastic deformation. By solving for a simplified version of the heat equation, we can estimate the strength of alumina during dynamic release.

The equation governing the temperature evolution of an adiabatically released material is

$$\frac{dT}{dt} = T\gamma : \frac{d\epsilon^e}{dt} + \frac{1}{c_v} \left[f_{TQ} \left(\sigma : \frac{d\epsilon^p}{dt} \right) \right] (4)$$

where ϵ^{e} and ϵ^{p} are the elastic and plastic strain tensors, respectively, σ is the Cauchy stress tensor, $\gamma = V(\partial \sigma / \partial E)_{V}$ is the Grüneisen tensor, c_{v} is the volumetric heat capacity, and f_{TQ} is the Taylor-Quinney factor. The former term represents the thermoelastic effect whereby reversible changes in density raise or lower the temperature of the material; this term is the origin of the isentropic cooling caused by release to ambient pressure in the absence of heating. The latter term represents plastic-work heating, which, in energetic terms, is simply the total plastic work rate $\sigma: d\epsilon^{p}/dt$ multiplied by the phenomenological Taylor-Quinney factor f_{TQ} , which expresses the fraction of plastic work dissipated as heat. The final release temperature is controlled by the relative magnitude of the above cooling and heating terms; the plastic-work heating is, in turn, controlled by the strength of alumina.

Using the heat equation to constrain the material strength accurately is extremely challenging for even the most intensively studied materials thanks to the unknown represented by the Taylor-Quinney factor. The fraction of plastic work converted to heat typically takes values between 0.70 and 0.95 in manufacturing forming processes (Taylor and Quinney, 1934,

pp.307-326), but there is growing evidence that f_{TQ} can take values beyond this range in extraordinary deformation scenarios: Taylor-Quinney factors of 0.5 or less were inferred in diamond undergoing ramp compression to 2 TPa (Lazicki et al, 2021, pp.532-535); meanwhile, values of f_{TQ} greater than unity are possible during release, due to additional heating from defect annihilation processes (Heighway et al, 2019, p.245501). In view of this considerable uncertainty, we present an approximate solution to the heat equation that suffices to give an order-of-magnitude estimate of the strength. The equation (whose derivation is given in the Supplemental Materials) reads

$$\frac{4}{3}f_{TQ}\left(\frac{\tau}{c_v^T s}\right)\frac{1}{\gamma} = \frac{\frac{T_R}{T_s} - \left(\frac{V}{V_0}\right)^{\gamma}}{1 - \left(\frac{V}{V_0}\right)^{\gamma}}, \quad (5)$$

where τ is the conventional material strength (i.e. $|\sigma_L - \sigma_T|/2$, where σ_L and σ_T are the stresses along the loading and transverse directions, respectively), V/V_0 is the compression ratio, and T_R/T_s is the ratio of the release temperature to the shock temperature. This equation allows us to gauge the magnitude of τ given alumina's EOS, its shock and release temperatures, and its Taylor-Quinney factor.

To estimate τ , we took the volumetric heat capacity c_v to be the classical value predicted by the Dulong-Petit law $(3nk_B)$. Given the paucity of data constraining the Taylor-Quinney factor in this regime, we further assumed that $f_{TQ} = 1$. With these assumptions, the heat equation yields a range from 5 GPa to above 9 GPa for the dynamic strength of alumina during release. The corresponding fitted release paths are shown in Fig. 5 for comparison with the isentropic release path. A dynamic strength of 5 GPa is in fact comparable to the strength of 5.2 GPa exhibited by alumina during *compression* to pressures just below the shock overdrive pressure ($\sigma_L \sim 15$ GPa), but exceeds the static strength of alumina in the steady shock state (around 2.5 GPa) (Bourne, 2007, p.073514). Our analysis demonstrates that the strength of dynamically releasing micron-scale alumina can easily reach the gigapascal level; stronger constraints upon the Taylor-Quinney factor in this regime would be needed to construct more accurate upper and lower limits on our strength estimate.

We should note that in addition to plastic work, a small amount of heat may be either supplied or absorbed by a phase transition from the $Rh_2O_3(II)$ -type phase back to the α -corundum phase. A discontinuity in the actual isentrope at the phase transition could also contribute to higher temperatures. The SESAME EOS and our calculations do not account for the phase transition, which has been reported to have a -3% volume discontinuity (Lin et al, 2004, pp.389-393; Mashimo et al, 2000, pp.2021-2024). In the future, obtaining XRD patterns after full release in material shocked to below the phase transition would help to compare and quantify its effect on the release dynamics.

Run #	X-ray timing	a (Å)	<i>c</i> (Å)	c/a	ρ (g/cm ³)	T (K)
269	9 ns	4.835±0.001	13.201±0.006	2.730±0.002	3.801±0.003	≥2060
252	10 ns	4.816±0.001	13.102±0.007	2.720±0.002	3.858±0.004	≥1580
270	11 ns	4.811±0.001	13.151±0.004	2.734±0.001	3.853±0.002	≥1630
273	13 ns	4.817±0.001	13.174±0.005	2.735±0.002	3.837±0.003	≤1770
Ambient	-	4.757	12.993	2.731	3.989	300

Table 4: Nominal temperature calculated from density after release from ~110 GPa shock (dual-beam drive). This calculation suggests a rough lower bound of 1580 K on the release temperature.



Fig. 5: Deviation from isentropic release in dual-beam runs. After release from shock, temperatures calculated from density are much higher than that predicted by isentropic release. The magenta triangles indicate nominal temperatures from the released a-corundum in our dual beam runs. The black dashed line shows our lower bound estimate of the release path, while the region in blue shows a range of possible release paths, for which we do not have a solid upper bound. This calculation assumes the dynamic strength remains constant throughout release.

5. Conclusions

In this study, we present the first *in situ* measurements of crystal structure in shock compressed alumina. We find that alumina shocked to 107 ± 8 GPa produces X-ray diffraction patterns consistent with the Rh₂O₃(II) phase, in agreement with the results of static and computational studies. This could potentially allow for the characterization of additional phase transformations in alumina at higher pressures that cannot be reached with static compression techniques, such as the proposed transition from the post-perovskite structure (Cmcm) to a U_2S_3 -type structure (Pmcn) phase transition that has been predicted to take place at ~370 GPa (Umemoto & Wentzcovitch, 2008, pp.6526-6530). We also utilize a relatively new method to estimate material strength from X-ray diffraction, and show its first application in an ionic solid.

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Fig. S1: X-ray diffraction images from all runs. Lower stress, single-beam shots are shown along the left, while higher stress, dual-beam shots are shown along the right. Only the largest detector, CSPAD0, is shown.



Fig. S2: Linear thermal expansion measurements were converted to density and plotted in black ((Watchman Scuderi & Cleek 1962, Hahn 1978, Schauer 1964, Amatuni & Shevchenko 1976, Yates Cooper & Pojur 1972, Fitzer & Weisenburger 1972, Ebert & Tingwald 1936, Mauer & Bolz 1958, Campbell & Grain 1961, Kompan et al 1999, Kostanovskiy et. al 2017.) The points were fit to a model of thermal expansion detailed in Watchman et al. 1962. Observed densities upon release from shock (dual-beam drive) are shown in red alongside the fit shown blue.



Fig. S3: Laser power of the drive over time in a) the single beam shots (~50 GPa) and b) the dual beam shots (~100 GPa).

Structure	a/b	a/c
Rh ₂ O ₃ (II) (this study)	1.540	1.362
Rh ₂ O ₃ (II) (Lin et al., 2004)	1.466	1.407
Perovskite (this study)	1.413	1.427
Perovskite (Lin et al., 2004)	1.407	1.466

Table S1: Ratios between lattice parameters of the high pressure phase found in this study at 107 ± 9 GPa on the Hugoniot, compared with the measurement made by Lin et al., 2004 at 113 GPa and 300 K.^[4]

Phase	Reflection (hkl)	Center (20)	Peak height (arb. units)	Peak area	FWHM (20)
	012	25.6374	77.0216	36.3124	0.442694
	104	35.2375	274.8	145.661	0.497723
	110	37.8675	190.101	87.3923	0.431667
	113	43.4608	647.403	264.938	0.384266
	024	52.6848	322.137	129.695	0.378045
	116	57.6504	681.732	258.89	0.356586
Unshocked a-cor	018	61.4689	84.486	29.2347	0.32492
	012	25.2313	86.929	63.8946	0.690177
	104	34.6678	328.635	282.477	0.807104
	110	37.2583	247.074	182.44	0.693353
	113	42.7506	494.463	397.632	0.755108
	024	51.8022	365.392	546.379	1.40409
	116	56.6658	473.842	637.466	1.26324
Released a-cor	018	60.4001	251.568	402.63	1.50284
	200	27.6743	57.1762	60.3495	0.991104
	111	31.3645	90.5588	177.583	1.84133
	002	39.5532	566.156	1297.55	2.15203
	211	39.8187	695.132	768.763	1.03845
	020	40.6364	177.808	222.22	1.17353
	021	45.4431	256.408	454.669	1.66504
	310	46.9748	1191.45	2519.4	1.98556
	121	47.6996	48.2467	109.37	2.12859
	202	48.9562	61.1808	50.3262	0.772397
	311	51.3121	135.04	279.219	1.94153
	212	53.391	164.734	271.778	1.54914
	400	57.1522	261.52	697.594	2.50472
Rh ₂ O ₃ (II)	022	58.0019	550.746	1294.73	2.20744

Table S2: Peak positions for run 269 (dual-beam, 9 ns X-ray timing) fit using the Powder Diffraction Tool in Fityk. Peaks were selected by structure factors from atomic positions determined by Lin et al (2004, pp.389-393).



Fig. S4: Fitted peaks in Fityk (run 269, dual-beam, 9 ns X-ray timing). Peaks assigned to the $Rh_2O_3(II)$ phase are shown in red, and those from a-corundum are shown in light blue (ambient pre-shock) and dark blue (released). The yellow line shows the sum of the fitted peaks, and the green dots show the data.

Peak Fitting

Azimuthally averaged lineouts from HEXRD were imported into the Fityk software. Background was subtracted using a spline function prior to fitting. Peak functions for each reflection were defined using the Powder Diffraction Tool were selected by their structure factors (based on atomic positions determined experimentally by Lin et al (2004, pp.389-393)), with some lower structure factor peaks being selected if they appeared clearly in the diffraction pattern. Peaks widths and heights were independent for each peak, and the centers were defined by the Bragg condition. Since the material was textured, it was not possible to refine the fit incorporating structure factors. Peak parameters were refined using the Levenberg-Marquardt method, a nonlinear least-squares fitting algorithm. Uncertainties in lattice parameters were estimated by Fityk and propagated to determine the uncertainty in density.



Fig. S5: Model of lateral release and probed region. Schematic of shock propagation and release in alumina as viewed from above, shown to scale. The diagrams show a snapshot in time of the different density states within the material. The X-ray beam, shown in blue, is angled 24° and centered on the surface of the polyimide ablator. The spatial extent of the shock wave is assumed to diminish at a 45° angle with distance in polyimide, and then at a 39° angle in alumina, calculated using the relation (green dashed lines). The dark red region indicates high density, shock-compressed material, while the light blue region shows material at ambient density and the purple region shows laterally equilibrated material below ambient density.

Lateral Release

In the XRD images collected at 9, 10 and 11 ns during dual-beam shots (runs 269, 252 and 270), we find diffraction peaks from both compressed and released material together in the same image. The timings of these observations, particularly at 9 and 10 ns, are not consistent with our estimates of timing for partial release from the kapton-alumina interface and shock break out from the rear free surface (Table 2). It is notable that this observation is limited only to shots conducted after the X-ray spot diameter was increased from 20 μ m to 80 μ m.

Since we have ruled out release from the rear surface due to shock break out based on the presence of ambient diffraction peaks, the most likely source for these peaks is lateral release, in which the area of the propagating shock wave diminishes with time and distance into the target, equilibrating with the surrounding unshocked material to a low but nonzero pressure state. This equilibrated intermediate region grows over time in both directions, via lateral shock propagating

outward as well as release waves propagating inwards. We believe that the increased spot diameter of 80 μ m allowed us to access release states that otherwise would not have been illumined. A two-dimensional schematic illustrating this interpretation is shown in Fig. S5.

Over time, the spatial extent of planarity of the compression wave is reduced due to the encroachment of release waves originating from the edge of the shocked region. Recent calculations suggest that the angle at which the shock front diminishes can be as high as 60° in a polyimide ablator at 80 GPa (Swift et al, 2008, p.066115). This angle can be calculated using the following equation detailed in Swift et al (2008, p.066115):

$$\phi = \arctan\left(\frac{\sqrt{c^2 + (U_s - U_a)^2}}{U_s}\right) (6)$$

where U_s is the shock velocity, c is the sound speed, and U_a is the ablation rate. Sound speed can be calculated using the following equation:

$$c = \frac{Y}{\rho}.(7)$$

Y, the Young's modulus is given by $Y = 3K(1 - 2\mu)$, where μ is Poisson's ratio and K is the bulk modulus.

The bulk modulus of alumina in the $Rh_2O_3(II)$ phase at 107 GPa was approximated using the Voigt-Reuss-Hill approximation from compliance tensor values calculated at high pressure by Duan, Karki and Wentzcovitch (1999, pp.1961-1966). Due to lack of constraint over alumina's Poisson's ratio at high pressure, the ambient value was used. With these assumptions, we determined ϕ to be ~39° in alumina. The ablation rate of polyimide in our experiments was unknown so we could not calculate ϕ , and instead used a widely accepted value of 45°.

In an attempt to simulate the shock propagation and release over space and time, we ran hydrodynamic simulations in HYADES to model both the single-beam and dual-beam shots (Fig. S6). However, the timings of partial release from the kapton-alumina interface and shock break out from the rear surface do not match up with the constraints determined using diffraction patterns. The reduction in spatial planarity of the shock front over time is a possible explanation for why shock breakout and full decompression occur closer together in time than predicted by HYADES simulations, which are one-dimensional and do not account for lateral release. Another possibility for the discrepancy is the failure of the simulation to take into account the strength of alumina, which has been found to remain high up to 100 GPa under shock compression (Kleiser, Chhabildas & Reinhart, 2011, pp.473-479.)

These experiments were part of one of the first investigations of shock-compressed minerals at the MEC. At this time, we were still learning how to create a well-defined laser spot and the optimal X-ray spot size to use. We did not use phase plates in the shots discussed here, and our exact laser spot diameter is not well constrained—it could range from 200 to 250 μ m, and the spot is not exactly circular. The uncertainty in laser spot size, along with the reduction in spatial planarity of the shock front over time, is a possible explanation for why our observed shock entry into alumina, break out from the rear surface and full decompression occur closer together in time than predicted by HYADES simulations.



Fig. S6: HYADES simulations. The shock propagation was modeled with HYADES, a 1-dimensional hydrodynamic simulation software. (A) Simulation of the single-beam shots. A 27 GPa pressure input is applied to the kapton (polyimide), yielding a pressure of 50 GPa in the alumina. The shock-wave enters alumina at 7.5 ns, begins releasing from the kapton-alumina interface at 12.5 ns and breaks out of the rear free surface at 13.9 ns. (B) Simulation of the dual-beam shots. A 42 GPa pressure input is applied to the kapton, yielding a pressure of 100 GPa in the alumina. The shock-wave enters alumina at 6.8 ns, begins releasing from the kapton-alumina interface at 10.1 ns and breaks out of the rear free surface at 12.3 ns. Timings at which X-ray diffraction images were taken are shown in red. Predicted shock breakout timings are not consistent with our XRD measurements (see Table 2).

Solution to simplified heat equation

In this section, we outline how we solved a simplified version of the heat equation in order to generate an estimate of the dynamic strength of alumina during release. We begin with the heat equation governing the temperature evolution of a dynamically releasing material element under adiabatic conditions:

$$\frac{dT}{dt} = T\gamma : \frac{d\epsilon^{e}}{dt} + \frac{1}{c_{v}} \left[f_{TQ} \left(\sigma : \frac{d\epsilon^{p}}{dt} \right) \right], (8)$$

where compressive strains are taken to be positive. The first term encodes the thermoelastic cooling experienced by the material due to its density decrease during release, while the second term expresses the plastic-work heating owed to plastic deformation. It is assumed that appreciable thermal conduction cannot take place over the nanosecond timescale of our dynamic-compression experiment, hence there being no diffusive term in the above equation. Our goal is to recast the heat equation into a form that can be solved analytically by making simplifying assumptions about the nature of the release process.

The first assumptions we make pertain to the symmetries of the material. We will assume that the Grüneisen tensor γ acts as a scalar (i.e. the thermal stress increases at an equal rate in all directions) and that the stress and strain tensors σ , ϵ^e and ϵ^p take the form diag(M_T, M_T, M_L), where M_L and M_T are the normal stresses or strains along the loading and transverse directions, respectively. The latter assumption essentially assumes that the crystal is isotropic in the plane transverse to the shock and that plasticity-induced rotation and shear do not introduce appreciable off-diagonal stresses in the laboratory frame. These conditions are almost certainly violated to some extent at the local level in a highly anisotropic crystal structure like alumina, but they may serve as a reasonable first-order approximation of the true material behavior in the absence of a more detailed model of plasticity/fracture in alumina. With these simplifying assumptions, the heat equation takes the form

$$\frac{dT}{dt} = T\gamma \left(2\frac{d\epsilon_T^{e}}{dt} + \frac{d\epsilon_L^{e}}{dt} \right) + \frac{f_{TQ}}{c_v} \left(2\sigma_T \frac{d\epsilon_T^{p}}{dt} + \sigma_L \frac{d\epsilon_L^{p}}{dt} \right).$$
(9)

The second set of assumptions expresses kinematic constraints the crystal must obey that reduce the number of strains that are allowed to vary independently. First, we invoke the assumption of *zero plastic dilatation*, which states that the irreversible part of the deformation

suffered by alumina does not, in and of itself, alter its volume. Mathematically, the constraint reads

$$2d\epsilon_T^{\ p} + d\epsilon_L^{\ p} = 0. (10)$$

This assertion is reasonable provided that the irreversible part of the deformation does not involve the creation of copious numbers of highly dilatational defects, like vacancies or voids. Second, we assume that the total deformation suffered by the alumina is purely uniaxial. That is, the total dimension of the sample cannot change in the directions transverse to the shock due to the exceptionally short timescale of the experiment. This tells us that

$$d\epsilon_T^{\ p} + d\epsilon_T^{\ e} = 0. \, (11)$$

There is reason to expect that this condition may be locally violated to some extent due to grain-grain interactions in polycrystalline alumina, but it will serve as a convenient approximation in the absence of a detailed model about such interactions. With the two kinematic constraints above, it is possible to recast the heat equation into the form

$$\frac{dT}{dt} = T\gamma \left(2\frac{d\epsilon_T^{e}}{dt} + \frac{d\epsilon_L^{e}}{dt} \right) - \frac{4f_{TQ}\tau}{c_v} \frac{d\epsilon_T^{e}}{dt}, (12)$$

where $\tau = (\sigma_T - \sigma_L)/2$ is the strength during release. The final simplifying assumption we make is that for the majority of the release process, the change in longitudinal elastic strain follows closely the change in transverse elastic strain (their difference essentially being set by the approximately constant dynamic strength of the material). Hence

$$\frac{dT}{dt} = 3T\gamma \frac{d\epsilon_T^{\ e}}{dt} - \frac{4f_{TQ}\tau}{c_v} \frac{d\epsilon_T^{\ e}}{dt}.$$
 (13)

This last assumption will fail particularly near the beginning of the release process, where some elastic behavior is expected, but, again, suffices for our simplified model.

The equation above can be trivially rearranged into a first-order ordinary differential equation that can be solved using the standard integrating factor:

$$\frac{dT}{d\epsilon_{T}^{e}} - 3\gamma T = -\frac{4f_{TQ}\tau}{c_{v}}.$$
 (14)

If we assume for simplicity that γ , f_{TQ} , τ , and c_v are approximately constant over the release process, the explicit solution to this equation is

$$T_{R} = [R + (1 - R)e^{-3\gamma\epsilon}]T_{S'} (15)$$

where T_s and T_R are the shock and release temperatures, ϵ is the linear elastic strain upon compression, and the dimensionless figure of merit

$$R = \frac{4}{3} f_{TQ} \left(\frac{\tau}{c_v T_s} \right) \frac{1}{\gamma} (16)$$

characterizes the ratio of the plastic-work heating to the thermoelastic cooling. Finally, by noting that to leading order, the quantity $e^{-3\epsilon}$ is equal to the volume ratio V/V_0 , we can rearrange the above into a form that allows easy evaluation of τ :

$$\frac{\frac{4}{3}f_{TQ}\left(\frac{\tau}{c_v T_s}\right)\frac{1}{\gamma} = \frac{\frac{T_R}{T_s} - \left(\frac{V}{V_0}\right)^{\gamma}}{1 - \left(\frac{V}{V_0}\right)^{\gamma}}.$$
 (17)

Note that if the material has zero strength ($\tau = 0$), the solution reduces to the familiar form

$$\frac{T_{R}}{T_{S}} = \left(\frac{V}{V_{0}}\right)^{\gamma} (18)$$

expected of a purely isentropic release.